PURE AND APPLIED CHEMISTRY

Official Journal of the International Union of Pure and Applied Chemistry

Executive Editor

P.D. Gujral, IUPAC Secretariat, Bank Court Chambers, 2-3 Pound Way, Templars Square, Cowley, Oxford OX4 3YF, England

IUPAC COMMITTEE ON PUBLICATIONS

E.D. Becker (USA) Chairman

D.H.M. Bowen (USA)
D.V.S. Jain (India)
A. Lawson (Germany)

H. Ratajczak (Poland) C.J.H. Schutte (RSA) K.I. Zamaraev (USSR)

Pure and Applied Chemistry Editorial Advisory Board

D.H. Whiffen Chairman

J.Haber (Poland)

Physical Chemistry Divsion
E. Fluck (Germany)

Inorganic Chemistry Division

N.J. Leonard (USA)

Organic Chemistry Division

A.D. Jenkins (UK)

Macromolecular Division

G. den Boef (Netherlands)

Analytical Chemistry Division

L.E. Coles (UK)

Applied Chemistry Division

W.C. Purdy (Canada)

Clinical Chemistry Division

E.D. Becker (USA)

Committee on Publications

J. Reedijk (Netherlands)

Interdivisional Committee on Nomenclature and Symbols

Since 1960 the Union has made available to chemists everywhere a very large amount of important material which it handles each year through its regular publication in *Pure and Applied Chemistry*. The objectives of the journal are:

- to publish the main invited lectures of symposia sponsored by IUPAC at which authoritative and up-to-date accounts of the subject of the symposium are presented by prominent scientists; the lecturers cover their own recent work and review the worldwide literature on the subject;
- to publish the recommendations of the Union's commissions on nomenclature, symbols and units;
- to publish technical reports on standardization, recommended procedures, collaborative studies, data compilations, etc.;
- to publish critically evaluated state-of-the-art commissioned review articles on important topics.

Subscription Information

Pure and Applied Chemistry is published monthly. Subscription rates for 1991 are £445.00 (UK), £475.00 (overseas) and US\$715.00 (USA & Canada) post free.

Subscriptions and free specimen copies are available from:



Blackwell Scientific Publications

-JOURNALS-

Osney Mead, Oxford OX2 0EL Tel: (0865) 240201

Lancaster Catalogue

89/90

UNITED KINGDOM Lancaster Synthesis Ltd. Eastgate, White Lund.

Morecambe, Lancs, LA3 3DY Freetone: 0800-262336 Telephone: 0524-36101 Fax: 0524-39727 Telex: 65151 (LNCSYN G)

U.S.A. AND CANADA Lancaster Synthesis Ltd.

P.O. Box 1000. Windham. New Hampshire 03087 Toll-free lines: 800-238-2324 Telephone: 603-889-3306 Fax: 603-889-3326

FRANCE

Lancaster Synthesis Ltd. 15 rue de l'Atome Z.I.. 67800 Bischheim. Strasbourg. Telephone: 88-62-26-90 Fax: 88-62-26-81 Telex: 870551 (LNCSTRM F)

GERMANY

Deutsche MTM Chemie GmbH. Laemmerspieler Straße 100A. 6052 Muhlheim am Main, Telephone: 06108 73 019 Fax: 06108 74 814

JAPAN

Hydrus Chemical Inc.
Tomitaka Bldg. 8-1.
Uchikanda 2-chome.
Chiyoda-ku,
Tokyo 101.
Telephone: (03) 258-5031
Fax: (03) 258-6535
Telex: 2324032 (HYDRUS J)

Completely revised 1152 pages 1000 new items

> 6500 literature references to some 2000 items Illustrated by 1500 reaction flow-charts Semi-bulk and bulk quantity indications Extensive cross-referencing

de, is converted to the Mannich reagent, N,N-oacetate, an excellent reagent for the α -dimethyl-Bull.Soc.Chim.Fr., 2707 (1970). Compare dimethyl-0131, p.423:

Me2NC=CH2 CF3COO

CH₂NMe₂

Deprotonation by lithium diisopropylamide at low temperature gives the unstable azomethine ylide, which undergoes 1,3-dipolar addition even with unactivated alkenes, to give pyrrolidines: *J.Chem.Soc.*, *Chem.Commun.*, 31 (1983):

Me₃NO

LDA, THF

 $H_2C = N CH_3$

MeN

42%

Compare also N-methylmorpholine-N-oxide, 5957, p.710.

Nickel acetylacetonate hydrate

[Nickel(II)2,4-pentanedionate hydrate] F.W. 274.94, m.p. ca 285 (dec), [3264-82-2] HARMFUL / POSSIBLE CARCINOGEN CH₃

50g 7.60 250g 30.40

£

Please ask for bulk prices (5Kg to 100Kg+)
Catalyst for a variety of useful coupling reactions, including:

Catalyst for a variety of useful coupling reactions, including:
Conjugate addition of alkynylaluminium reagents to enones: *J.Am.Chem.Soc.*, **100**, 2244 (1978):

r°.

RC ≡ CAIMe₂

Ni(acac)₂

RC **≡** C

Conjugate addition of cis-alkenylzirconium reagents, from the hydrozirconation of alkynes, to Michael acceptors, with retention of configuration: *J.Am.Chem.Soc.*, **102**, 1333 (1980).

Coupling of Grignard reagents to give biaryls: *J.Org.Chem.*, **41**, 2252 (1976).

Coupling of Grignard reagents with silyl enol ethers of both aldehyles and ketones, to give alkenes. In contrast to dichlorobis(triphenylphosphine)nicke reagent gives the thermodynamically more stable alkene: *Tetrah* (1980):

R OSiMe₃

PhMgBr

Ni(acac)₂

p F

SEND NOW

MAKE SURE YOU HAVE
YOUR PERSONAL CORY

LANCASTER SYNTHESIS - PART OF MTM RESEARCH CHEMICALS